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A NEW SYNTHESIS OF α, β -unsaturated γ - and δ -lactones VIA intramolecular acylation of α -sulfingl carbanion

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Summary: A new synthesis of α , β -unsaturated γ - and δ -lactones involving the intramolecular acylation of α -sulfinyl carbanion followed by pyrolysis is described.

 α,β -Unsaturated γ - and δ -lactones are key structural subunits of natural products^{1,2} and valuable synthetic intermediates.^{2,3,7} As a consequence of their importance, many methods for the preparation of these compounds have been devised.⁴ However, exploration of new procedures for construction of such frameworks from readily available acyclic precursors is still required. In connection with our studies on the intramolecular acylation of α -sulfinyl carbanions in organic synthesis,⁵ we were interested in finding a new efficient and general method for preparation of α,β -unsaturated γ - and δ -lactones. Herein, we wish to describe a new synthesis of compounds <u>4</u> and <u>5</u> utilizing the intramolecular acylation of α -sulfinyl carbanion as shown in the following Scheme.



The cyclisation of the sulfoxide carbonate $\underline{1}^6$ to the α -phenylsulfinyl γ - and δ -lactones $\underline{2}$ and $\underline{3}$ was cleanly achieved by employing lithium diisopropylamide (LDA) in tetrahydrofuran (THF). Thus, treatment of $\underline{1}$ with LDA (2.0-2.2 equiv) in THF at -78° followed by warming to ambient temperature over 10 hr gave the cyclised products $\underline{2}$ and $\underline{3}$ in moderate to good yield after

quenching the mixture with glacial acetic acid followed by stirring at room temperature for 1 hr. Neat pyrolysis of the cyclised product 2 and 3 at 120° C under reduced pressure (0.05-0.1 torr) for 2 hr followed by preparative thin-layer chromatography (SiO₂) afforded the expected α , β -unsaturated γ - and δ -lactones in good yield. The results are summarized in Table 1. Compound 5b is massoia lactone, which is isolated from the bark oil of *Cryptocarya massoia*.^{7,8}

 <u>1</u>	<u>2</u> or <u>3</u> (%) ^{a,b}	$\frac{4}{2}$ or $\frac{5}{2}$ (%) ^{a,b}
<u>1</u> : $n = 1$		
la, R = H	<u>2a</u> (53)	<u>4a</u> (72)
<u>1b</u> , $R = CH_3CH_2$ -	<u>2b</u> (79)	<u>4b</u> (68)
<u>1c</u> , $R = n - CH_3(CH_2)_3 -$	<u>2c</u> (78)	<u>4c</u> (61)
<u>1d</u> , $R = n - CH_3 (CH_2)_5 -$	<u>2d</u> (80)	<u>4d</u> (76)
<u>le</u> , $R = n - CH_3 (CH_2)_6 -$	<u>2e</u> (72)	<u>4e</u> (58)
<u>lf</u> , $R = n - CH_3 (CH_2)_{10}$	<u>2f</u> (75)	<u>4f</u> (80)
$\underline{1}: n = 2$		
lh, R = H	<u>3a</u> (53)	<u>5a</u> (85)
\underline{li} , R = Ph	<u>3b</u> (68)	<u>5b</u> (96)
<u>lj</u> , $R = n - CH_3(CH_2)_3 -$	<u>3c</u> (85)	<u>5c</u> (86)
<u>lk</u> , $R = n - CH_3 (CH_2)_4 -$	<u>3d</u> (55)	<u>5d</u> (80)
<u>11</u> , $R = n - CH_3(CH_2)_5 -$	<u>3e</u> (87)	<u>5e</u> (85)
\underline{lm} , R = $n - CH_3(CH_2)_6$ -	<u>3f</u> (79)	<u>5f</u> (80)
$ln, R = n - CH_3(CH_2)_{10}$	<u>3g</u> (83)	<u>5g</u> (87)

Table 1: Preparation of Compounds 2,3,4 and 5.

a) All products have been characterised by spectral data.

b) Isolated yield after silica gel preparative thin-layer chromatography.

Furthermore, the cyclised products of types $\underline{2}$ and $\underline{3}$ appear to be important intermediates for the preparation of the saturated γ - and δ -lactones which are of much value as flavor substances and insect pheromones.⁹ Thus, treatment of the sulfoxides $\underline{3d}$ and $\underline{3g}$ with Al/Hg¹⁰ in aqueous THF furnished the δ -lactones <u>6a</u> and <u>6b</u> in good yield. The δ -lactones <u>6b</u> has been proposed to be the pheromone responsible for the social behavior of the oriental hornet, Vespa orientalis.¹¹



The above results clearly show that the present cyclisation is of general use for the preparation of α , β -unsaturated γ - and δ -lactones as well as the saturated ones.

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- The sulfoxide carbonates <u>1</u> were prepared in good overall yields according to the following equation (cf. V. Reutrakul, P. Tuchinda and K. Kusamran, <u>Chem.Lett.</u>, 1055 (1979).

$$RMgBr + PhSCH_2(CH_2)_nCHO \xrightarrow{1. THF/-78^{\circ} \rightarrow RT} R-CH(CH_2)_{n+1}SPh$$

$$1. ClCO_2Me/Pyridine$$

$$2. NalO_4/MeOH/H_2O$$

$$1 (n = 1, 2)$$

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